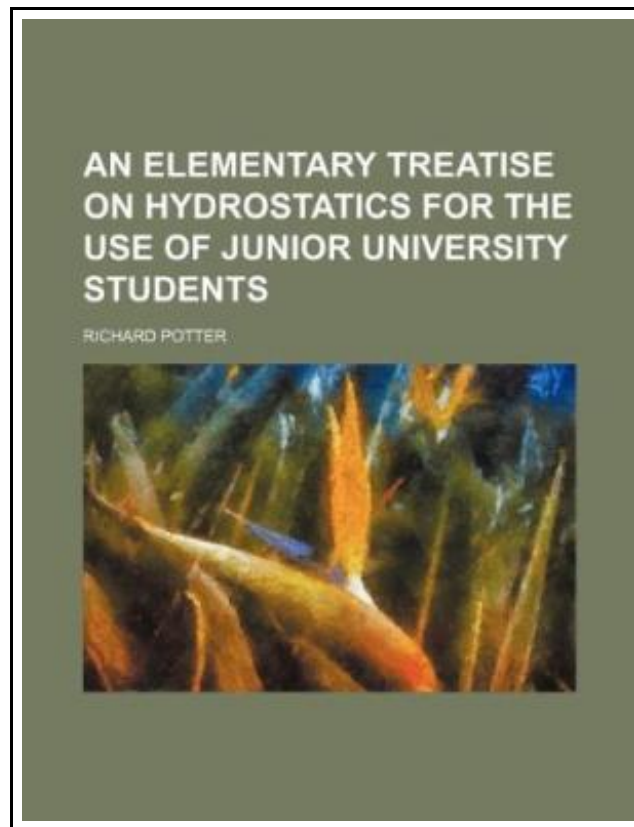


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1859 Excerpt: $p = l$ = the pressure to be applied to change the volume from F to V Applying the result of Prop. 30 to the first case, we have $e - e = 236 - 1 - 1$ ounces = 2623 ounces. In the second case $e - e = 2361 - l$ ounces = 22-13 pounds. Ex. 2. If the tube in the last question had been 30 inches long, shew that the results would have been respectively 1686 ounces and 4057 pounds. Ex. 3. If an air-bubble, which is a sphere of j th inch diameter at a depth of 102 feet in water, ascends to a depth of 68 feet, show that its diameter is then i th inch, and when it ascends to a depth of 34 feet its diameter is r th inch. $r = 794$ Ex. 4. If the volume of a gas is 100 cubic inches at the temperature 60 Fahrenheit, what will be its volume under the same pressure at 16 below the zero of Fahrenheit's scale? If V_0 be the volume at the freezing point, V 60 temperature, $V - 16$ then $F = F_0(l + a) = F_0(1 + F = F_0(l - ar) = F_0(l - 6)$; and since by dividing $y = +o$? $223 V = V - r = 8544$ cubic inches, 261 since $V = 100$ cubic inches. Ex. 5. If the density of the atmospheric air is called unity at the freezing point, what is its density at the boiling point of water? $p V$ Since generally $= 7$, also 100 measures of air at the freezing point become 1364 at...



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